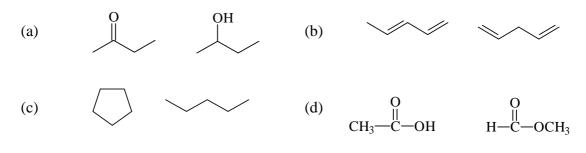


School of Chemistry

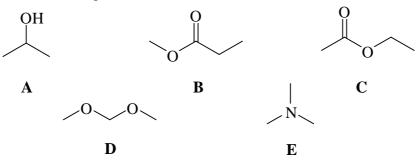
Chemistry 1B Advanced and SSP (CHEM1902/1904)

Problem Sheet 2

- 1. Low resolution mass spectrometry indicated a molecular ion at m/z = 58. The compound could be C₃H₆O or C₄H₁₀ or C₂H₆N₂. High resolution mass spectrometry gave a value of m/z = 58.0530. Using the following more accurate atomic weight data determine which compound it is. Data: ¹H 1.0078; ¹⁴N 14.0031; ¹²C 12.0000; ¹⁶O 15.9949
- 2. Which spectroscopic technique would most readily distinguish between the following pairs of compounds? Give a brief reason.



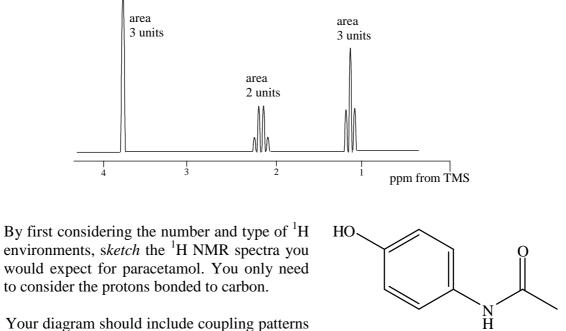
3. Consider the following molecules, **A** - **E**.



More than one answer may be correct. Give all correct answers.

- (a) Which of the molecules would give three signals in the ¹H NMR spectrum?
- (b) Which of the molecules would possess a ¹H NMR spectrum consisting of only one resonance?
- (c) Which of the molecules would possess a ¹H NMR spectrum consisting of two signals in the ratio 1:3?
- (d) How many singlets would be observed in the ¹H NMR spectrum of \mathbf{D} ?
- (e) Which of the molecules would possess a ¹H NMR spectrum containing a singlet, a triplet and a quartet signal?

(f) Examine the ¹H NMR spectrum below. To which of the compounds does it belong? (Hint: See page E35-11 of the Laboratory Handbook for approximate chemical shifts.)



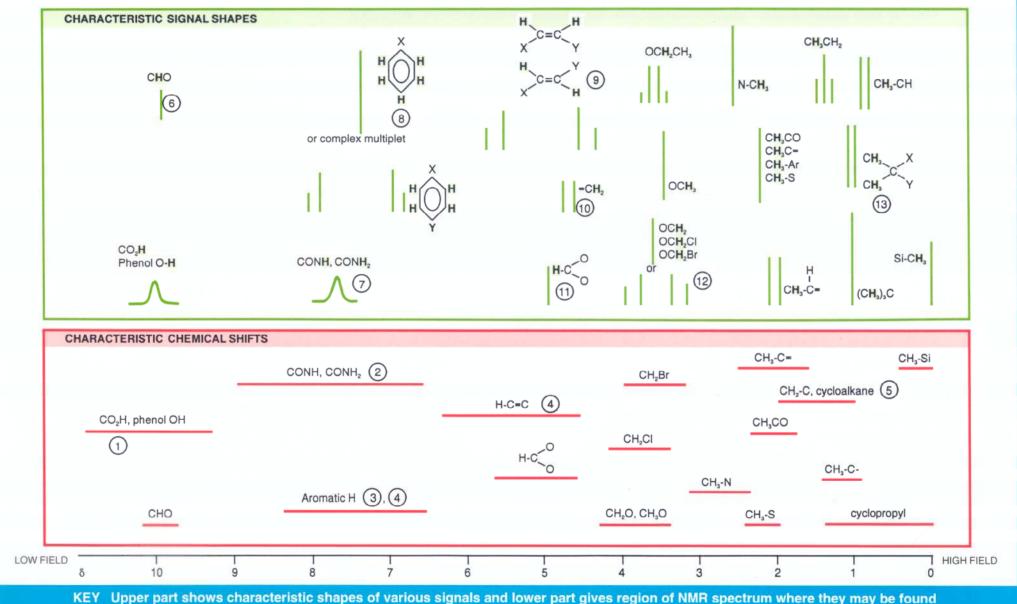
Your diagram should include coupling patterns and integration of each multiplet.

4.

paracetamol

(Hint: Use the table on the next page to find out the approximate chemical shifts.)

GlaxoWellcome 'H NMR Spectral Characteristics



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USE Use lower part for the possible origin of signals in your spectrum then refer to upper part for distinguishing features. Not in circles refer to key overleaf